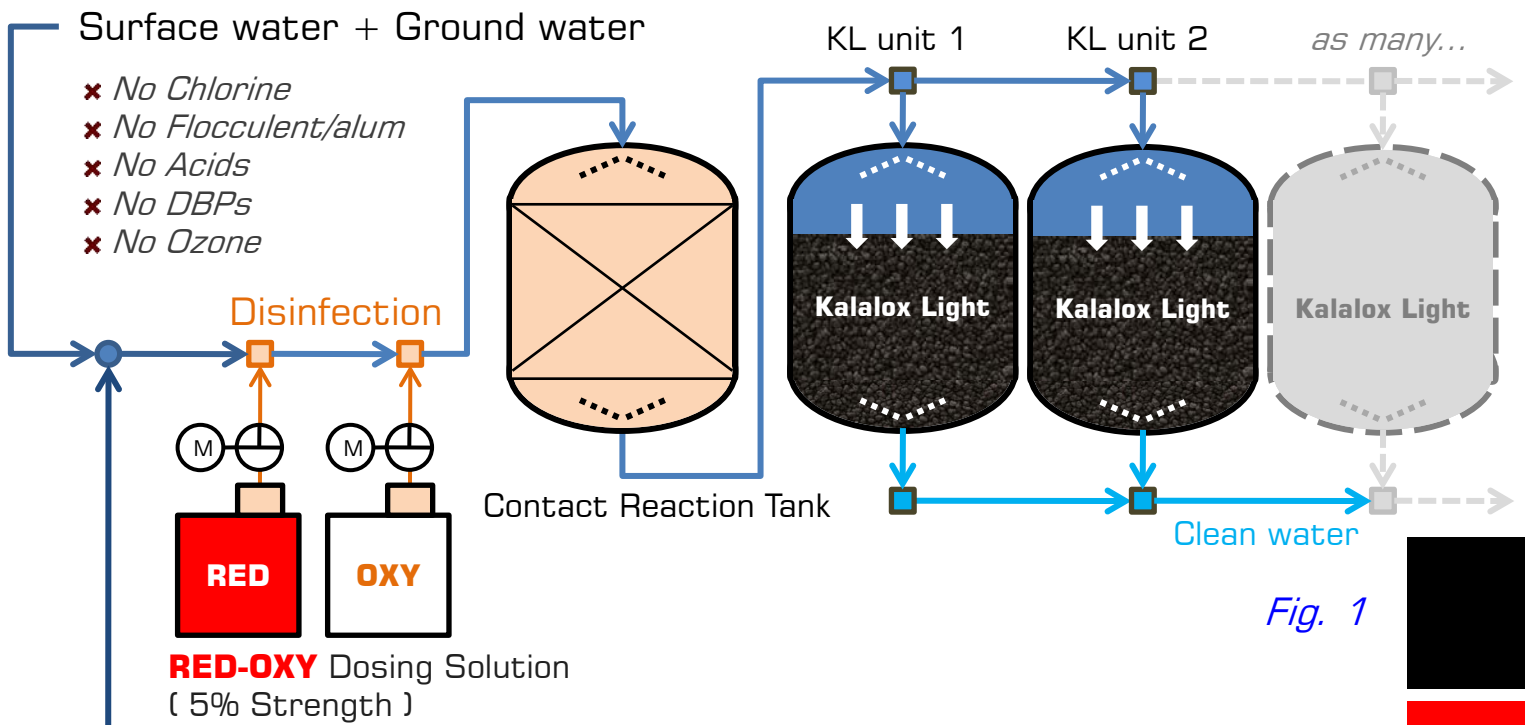


# RED-OXY<sup>®</sup>

## TREATMENT

### OXIDATION - ADSORPTION - FILTRATION



### Contaminants treatable by RED-OXY

Metal Cations removed	Anions removed	Oxidation Inorganic/organic Contaminants	Disinfectant Oxidant
Endocrine Disrupting Chemicals (EDCs)	Pharmaceuticals	Pesticides	

## Metal Cations Removed

Aluminum	Al (III)
Arsenic	As (III)
Barium	Ba (II)
Cadmium	Cd (II), Cd (III)
Calcium	Ca (II)
Cerium	Ce (III)
Cobalt	Co (II)
Copper	Cu (II)
Lead	Pb (II)
Magnesium	Mg (II)
Manganese	Mn (II)
Mercury	Hg (II)
Potassium	K (I)
Silver	Ag (I), Ag (II)
Thalium	Tl (III)
Tin	Sn (II)

## Anions Removed

Arsenate	As (III)
Arsenite	As (V)
Ammonia	NH <sub>3</sub>
Chromate	CrO <sub>4</sub> <sup>2-</sup>
Fluoride	F <sup>-</sup>
Molybdate	MoO <sub>4</sub> <sup>2-</sup>
Phosphate	PO <sub>4</sub> <sup>3-</sup>
Selenite	SeO <sub>3</sub> <sup>2-</sup>
Silicate	SiO <sub>2</sub> <sup>2-</sup>
Sulfate	SO <sub>4</sub> <sup>2-</sup>
Sulfite	SO <sub>3</sub> <sup>2-</sup>

## Oxidation

(inorganic & organic  
Contaminants)

1, - Diaminopropane  
1,2 - Ethanediol  
1,2 - Propanediol  
1,2,4 - Butanetriol  
1,3 - Propanediol  
2 - Mercaptobenzoic acid  
2 - Mercaptoethanesulfonic acid  
3 - Amino-1- propanol  
3- Mercaptopropionic acid  
Acetaldehyde  
Acetone  
Alpha-Hydroxy-toluene  
Ammonia  
Aniline  
Benzenesulfinate  
Chloral  
Cyanide  
Cysteine  
Cysine  
Diethylamine  
Diethylsulfide  
Dimethylamine  
Dimethylglycine  
Dimethylsulfoxide  
Ethyl alcohol  
Ethyl ether  
Ethylene glycol  
Ferrocyanide  
Fonic acid  
Formaldehyde  
Glycerol  
Glycine  
Glycoaldehyde  
Glycolic acid  
Glyoxal  
Glyoxylic acid  
Hydrazine  
Hydrogen sulfide  
Iminodiacetic acid  
Isopropyl alcohol  
Methionine  
Methyl alcohol  
Methylamine  
Methylhydrazine  
Neopentyl alcohol  
Nitriloacetic acid  
Nitrite  
Nitrosamines  
Methanol

## Oxidation

(Inorganic & organic  
Contaminants)

*Continues...*

N-methyliminodiacetic  
acid  
Oxylic acid  
Phenol  
p-Aminobenzoic acid  
P-Hydroquinone  
p-Nitroaniline  
p-Toluidine  
Sarcosine  
Thioacetamide  
Thiodiethanol  
Thiosulfate  
Thiourea  
Thioxane  
Trimethylaldehyde

## Disinfectant & Oxidants

Aerobic spore-bearers  
B. Cereus  
Bryopsis sp.  
Caulerpa taxifolia  
Dasya baillouviana  
Enteromorpha intestinalis  
Escherichia coli (E. Coli)  
F-specific RNA-coliphage QB  
f2 Coliphage  
S. aureus  
S. bovis  
S. globigii  
S. facalis  
S. flexneri  
Sphaerotilus  
S. Typhumurium  
Styela plicata  
Sulfite-reducing clostridia  
Thermotolerant coliforms  
Total coliform

1,1,2,2 – tetrachloroethane  
1,1,2 – trichloroethane  
1,1 – dichloroethane  
1,2 – dichlorobenzene  
1,2 – dichloroethylene  
1,2,3 – trichlorobenzene  
2 - Chlorophenol  
2- Nitrophenol  
2,4,6 – Trichlorophenol  
2,4 – Dichlorophenol  
Acenaphene  
Anthracene  
Bromodichloromethane  
COD  
Chlorobenzene  
Dichloromethane  
Diethylphthalate  
Dimethylphthalate  
Ethylbenzene  
Hexachlorobenzene  
Nitrobenzene  
Naphthalene  
Pentachlorophenol  
Phenanthrene  
Toluene  
Trichloroethylene

## Endocrine Disrupting Chemicals (EDCs)

Bisphenol A  
Estrone (E1)  
17 b-Estradiol (E2)  
17 a-Ethinylestradiol (EE2)  
16 a-Hydroxyestrone  
4-Nonylphenol  
4-tert-Octylphenol

## Pharmaceuticals

Sulfamethoxazole  
Ibuprofen

## Pesticides

- 2,4 – Dichlorophenoxyacetic acid
- 2,4,5 - Trichlorophenoxyacetic acid
- Dursban
- EDB (Ethylene di-bromide)

## High Purity FERRATE

**Red-Oxy®** process of mixing Hydrated ferric solution and strong OXYDES-P solution is the easiest method to produce **pure Ferrate** in the reaction tank. The **purity of Ferrate** is more than **99%** in the mixed form. The **Ferrate** reduced is an exclusive process of Watch-Water Germany.

**Red-Oxy®** is the safest oxidant, inexpensive and “environmental friendly”, especially for potable water and waste water treatment applications. **Red-Oxy®** is an ideal treatment for industrial and municipal effluent containing hazardous organic and inorganic compounds as explained on page no. 2-3. Using **Red-Oxy®** there is no need to dose poisonous and corrosive fesses like chlorine, hypochlorite or ozone. These oxidants have deleterious side effects. Additionally, the handling of chlorine, hypochlorite, HOCl, chlorine dioxide or ozone are potential danger to workers due to their high toxicity. And a major disadvantage of chlorine and chlorine dioxide or any other chlorine-containing oxidant produce, chloramines, chlorinated aromatics, chlorinated amines or hydrocarbons. All of these oxidants are potential mutagens or carcinogens, are for sure more toxic than the parent contaminants.

**Red-Oxy®** a new oxidant is designed to move away from chlorine, as well as ozone. Both of the compositions in red and oxy are oxidation products and 100% biodegradable. The **ferrate** molecule precipitates out of solution as  $Fe(OH)_3$  and now the adsorption process starts to collect cation as well as anions from the water. The iron containing sales can be easily filtered out by **Katalox-Light** leaving iron-free water containing innocuous by-products.

## Red-Oxy® Process

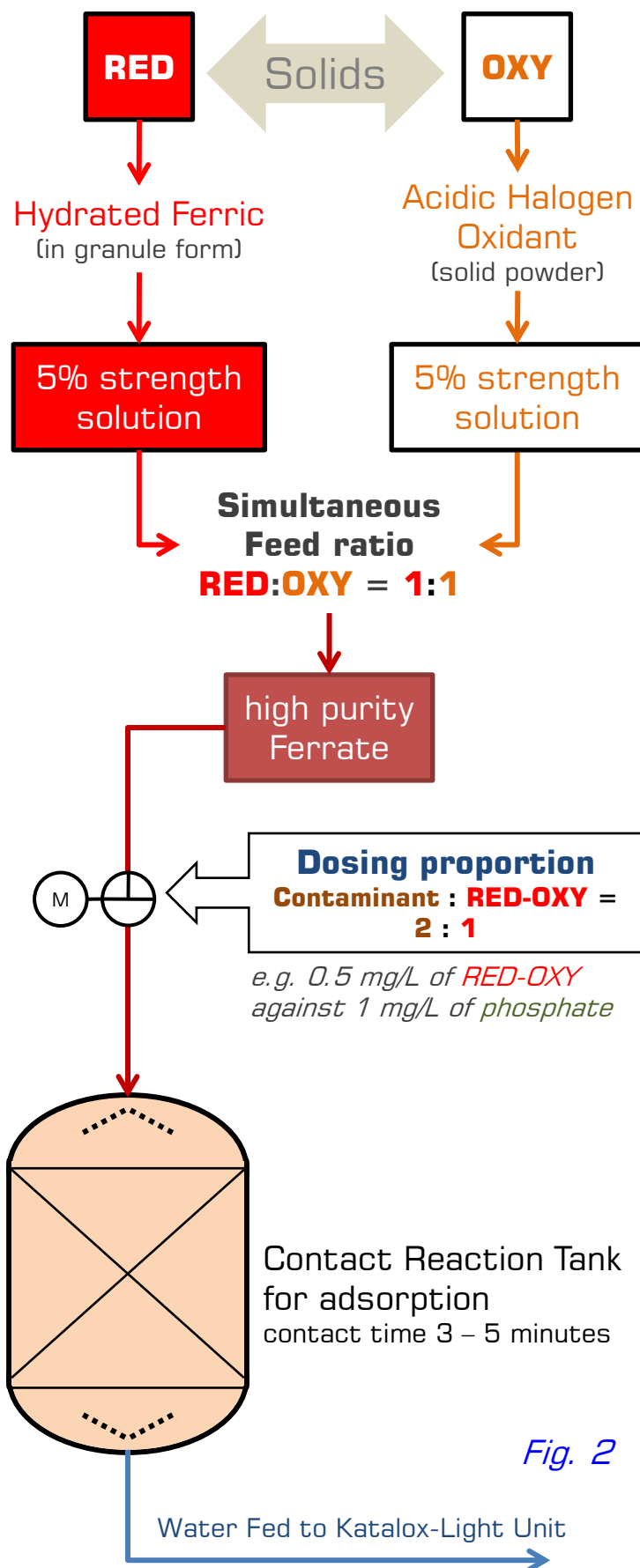


Fig. 2

## What is Red-Oxy® ?

Red-Oxy® Generated Ferrate with Watch-Water® process is the most stable form of **Ferrate** (VI) because it is generated with Ferric Hexahydrate granules. The oxidant used in this process is a strong acid halogen which generated high voltage of oxidant

without any DBPs. REDOX potential of the oxidant is as high as Hydroxyl Radicals. Watch-Water® has no doubt that its proprietary process will be commercialize world-wide with its own branches or through very close partners.

Watch-Water® understands chemistry. Generated Ferrate with its proprietary process of **ONLY TWO COMPONENTS** uses

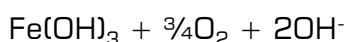
**Hexahydrate Granules** ←————→ **RED**

+

**Acidic Halogen Oxidant** ←————→ **OXY**

Red-Oxy® spontaneously decomposes in the presence of all contaminants listed on page 2 & 3, in any kind of water into strong oxygen and the most powerful Adsorbent based no ferric hydroxide with the surface area of 3500 m<sup>2</sup>/gram.

The chemical formation reaction is



This reaction is the strongest for the Oxidation-Adsorption of metals, non-metals and or organic contaminants in water and wastewater treatment. These include ammonia, cyanide, thiocyanate and very high concentration of hydrogen sulfide. All other contaminants are listed on Page 2 & 3. As many of the reactions are pH based reaction it can be controlled in the process with **OXY** (the most powerful oxidant and disinfectant against viruses and Coli-form Bacteria). Inactivation of viruses and all kind of bacteria occur faster as the pH drops, a phenomenon that has been attributed to mono-protonated form of  $\text{HFeO}_4^-$

Red-Oxy® treatment can be done without investing on the equipments as in most of the systems worldwide the dosing equipment can be used from existing feed pumps which will bring substantial improvements in finished water quality, especially as regards to trace organic contaminants and DBPs. Most water treatment systems, regardless of their size, use a coagulant which in future is **RED** and a chemical disinfectant which in future is **OXY** and the sand filter in future are **Katalox-Light** systems. However only adding ISOFT Corrosion Control chemicals may be necessary if after treatment corrosion-control is needed. Watch-Water has developed this technology in INSTANT form of **RED & OXY** to save transportation cost of chemicals.

Solutions for all contaminants problem just got much easier. How to approach come thousand so of communities, municipalities to use this innovative treatment : **Red-Oxy®** - Oxidation and Adsorption represents an improved Multiple Applications.

*All conditions* → *One Solution*

## Oxidant and Adsorption Equipments

### First : Dosing Equipment for **RED**

In order to get the best results and to ensure the necessary Ferric Hydroxide production, the operator should take into account the „Total Contaminants“ to be removed.

Total contaminants and dosing ratio is 1 to 1. The dosing tank with mixer and the dosing pump should make the solution for one week. The flow control equipment to dose should be proportional. Watch-Water recommends the use of water-meters with contact cables.

Example:	Cations (mg/L)	Anions (mg/L)
	Iron	As (III)
	Lead	As (V)
	copper	Phosphate
		silicate
	Value1	Value2

Total **B** = Value1 + Value2. Select dosing **B/2** amount of the prepared 5% dosing solution.  
Treatable water volume =  $200000 / B$  [m<sup>3</sup>]

e.g. to neutralize 80 mg/L of phosphate dose 40 mg/L of the dosing prepared solution.  
This would treat 2500 m<sup>3</sup> of water with dosing solution prepared from 5 kg of **Red-Oxy®**.

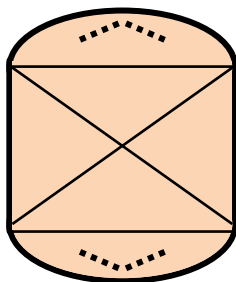
#### Note:

- If **ferrate** dosing is low the results may not satisfactory and on the other hand if ferrate yield is too high, the pressure drop in the **Katalox-Light** unit will be high.
- There is virtually no limits of BOD or COD including TOC.

### Second : Dosing Equipment for **OXY**

In order to achieve the best oxidation results to treat mixture of one or more impurities such as Biological impurities the same amount of oxidant to be used as in the process of Adsorbent production in the reaction tank. Biological impurities are those materials having biological origin. Thus any cells, bacteria, viruses, tissues or components thereof, whether from plants or animals can be considered to be biological impurities. This process will destroy sulfur-containing impurities and compounds containing sulfur atom including nitrogen-containing impurities, radioactive impurities etc. All other impurities as listed on the "contaminant list" (page 2 & 3).

### Third : Reaction Tank



Simple reaction tank should be chosen to provide a internal contact time of 3 to 5 minutes

Flow rate	Reaction Tank models	Inlet/Outlet	Contact time
m <sup>3</sup> /h		inches	minutes
0.5 – 1.0	10x54	¾	3 to 5
1.0 – 2.0	13x54	1	3 to 5
2.0 – 3.0	16x65	1	3 to 5
3.0 – 5.0	18x65	1½	3 to 5
5.0 – 8.5	24x69	2	3 to 5
8.5 – 13.0	30x72	2 to 2½	3 to 5
13.0 – 21.0	36x72	2 to 2½	3 to 5
22.0 – 36.5	48x82	3 to 4	3 to 5

## Katalox-Light Filtration

Water is fed to the **Katalox-Light** units for the filtration stage after Contact Reaction Tank. For **Katalox-Light** sizing please check **Katalox-Light** Technical Datasheet.

All salts and impurities captured by adsorbent are easier to filter out with **Katalox-Light** media, leaving iron free water containing none of the toxic byproducts.

In addition the nature of **Katalox-Light** can be utilized in urban or any industrial water

treatment plants. Since the **Red-Oxy®** technology is the highest effective and disinfectant technology, it is possible to replace every Chlorinated drinking water equipment.

Therefore, any water, wastewater, irrigation water, surface water or ground water mixed with organic, inorganic or biological impurities in water can install **Red-Oxy®** Adsorption/Disinfection equipment.

*Watch-Water® is proud to announce the newest addition to the  
Oxidation and Adsorption in One process, **Red-Oxy®** Treatment*

## Message from the Leader board

Our partners will build

- Laboratory pilot Scale
- On-site Pilot Scale

We will be using any kind of water samples.  
This will allow us to prove the technology.

If you ask, is **FeO<sub>4</sub><sup>2-</sup>** is the solution for

- Disinfection
- Oxidation and
- Adsorption followed by KL filtration  
- for water treatment in future ?

The answer is YES!

**Red-Oxy®** is the most powerful multi-purpose and environment friendly technology known in water-treatment.

**Red-Oxy®** is available as INSTANT product (solid granule/powder form) that can be delivered worldwide without unnecessary water.

99% purification/separation can be achieved using **Red-Oxy®** (Ferrate Hexahydrate) Technology.

Packaging:

- 4 x 5 kg bags of **RED** (Hydrated Iron in solid form) in a box
- 4 x 5 kg bags of **OXY** (oxidizer chemical in solid form) in a box

**Distributed by:**

Address:

Tel:  
Fax:  
Email:

**Manufactured by:**

**WATCHER®**

**Watch Water®, Germany**  
A Water Company

Address: **Fahrlachstraße 14  
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